

# How will Ocean Acidification Affect Baltic Sea Ecosystems? An Assessment of Plausible Impacts on Key Functional Groups

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**Abstract** Increasing partial pressure of atmospheric CO<sub>2</sub> is causing ocean pH to fall—a process known as ‘ocean acidification’. Scenario modeling suggests that ocean acidification in the Baltic Sea may cause a  $\leq 3$  times increase in acidity (reduction of 0.2–0.4 pH units) by the year 2100. The responses of most Baltic Sea organisms to ocean acidification are poorly understood. Available data suggest that most species and ecologically important groups in the Baltic Sea food web (phytoplankton, zooplankton, macrozoobenthos, cod and sprat) will be robust to the expected changes in pH. These conclusions come from (mostly) single-species and single-factor studies. Determining the emergent effects of ocean acidification on the ecosystem from such studies is problematic, yet very few studies have used multiple stressors and/or multiple trophic levels. There is an urgent need for more data from Baltic Sea populations, particularly from environmentally diverse regions and from controlled mesocosm experiments. In the absence of such information it is difficult to envision the likely effects of future ocean acidification on Baltic Sea species and ecosystems.

**Keywords** pH · CO<sub>2</sub> · Cod · Herring · *Mytilus* · Spring bloom · Phytoplankton · Skagerrak · Kattegat

## INTRODUCTION

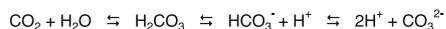
The open oceans act as a major buffer for atmospheric gases. Of the atmospheric CO<sub>2</sub> that has been released by mankind since the onset of industrialisation  $\sim 30\%$  has been absorbed by the oceans. This has caused the average pH of the open oceans to fall by 0.1 units (IPCC 2007)—a

seemingly small change, but one that is equivalent to a ca. 25 % increase in acidity. The underlying chemical processes are well understood (see Box), and consequently unlike climate scenarios, future changes in ocean pH can be predicted reliably from scenarios of future atmospheric CO<sub>2</sub> concentrations. Those scenarios project possible reductions in open ocean pH of as much as 0.35 units (equivalent to a 3 times increase in acidity) within the coming 100 years (Cao et al. 2007; IPCC 2007).

Closer to shore the effects of atmospheric CO<sub>2</sub> on coastal seawater pH are complicated by local effects of run-off, eutrophication, upwelling, atmospheric deposition and remineralisation, all of which also influence local biogeochemistry (Doney et al. 2007; Omstedt et al. 2009, 2010). Consequently, coastal pH is more variable and difficult to predict than that of the open ocean (Andersson et al. 2005). Data from coastal regions in the southern Baltic Sea reflect this: pH of the Kiel fjord varies by ca. 0.7 pH units seasonally (Thomsen et al. 2010), and diurnal fluctuations of  $\pm 0.15$  pH units are common in shallow bays of the Skagerrak (personal observation). Similar observations have been made in a variety of open ocean and coastal locations (Wootton et al. 2008; Hofmann et al. 2011). These diurnal changes—and a substantial fraction of the seasonal changes—in pH are driven by direct effects of photosynthesis and respiration. Until relatively recently such variability in seawater pH had been largely overlooked in ocean acidification literature, however, it is now clear that temporal variation in pH is a pervasive natural feature of marine systems. The effects of superimposing near-future ocean acidification on this variation are as yet unknown, but it seems clear that many marine organisms may already be adapted to more pH variability than was previously thought.

### Ocean Acidification – a simplified primer

When CO<sub>2</sub> dissolves in seawater it forms carbonic acid, which dissociates to form an equilibrium with hydrogen ions, bicarbonate ions and carbonate:



This equilibrium is dominated by bicarbonate, (CO<sub>2</sub> ≈ 1%, CO<sub>3</sub><sup>2-</sup> ≈ 8% and HCO<sub>3</sub><sup>-</sup> ≈ 91%, of total dissolved inorganic carbon, “DIC”).

Continued uptake of CO<sub>2</sub> by the oceans drives this equilibrium toward the centre, increasing the concentration of Hydrogen ions, and thereby reducing pH. This pH shift changes the equilibrium between bicarbonate and carbonate, driving that equilibrium also toward the centre, thereby depleting the available carbonate pool. This increases the solubility of CaCO<sub>3</sub>.

Increasing partial pressures of atmospheric CO<sub>2</sub> (pCO<sub>2</sub>) have already caused a ~30% increase in acidity (pH decline of 0.1 units) in the global oceans since the onset of the industrial revolution. “Business as usual” scenarios for the remainder of this century project that atmospheric CO<sub>2</sub> levels will continue to rise, leading to increased dissolution of CO<sub>2</sub> in the oceans, and further reductions in pH. This has been termed “ocean acidification”.

The capacity of seawater to buffer the addition of weak acids (such as H<sub>2</sub>CO<sub>3</sub>) is directly related to the total alkalinity, which is closely related to the salinity. Seawater in the Skagerrak–Kattegat–Baltic Sea system has reduced alkalinity and therefore reduced buffering capacity, with the result that pH is more variable (and generally lower) than in the open ocean. For example, annual pH range for surface waters in the central Kattegat (SMHI Station Anholt E) for the period 1992–2007 was 8.06–8.42 pH units, for the less saline central Baltic Sea (SMHI Station BY15) was 8.02–8.70 pH units, and for the very brackish northern Bothnian Bay (SMHI Station F9) was 7.40–8.37 pH units (SMHI 2011). Future increases in atmospheric CO<sub>2</sub> likely cause substantial shifts in pH in these waters, especially at lower salinities and alkalinities. Climate scenario projections for the end of the century suggest that surface water pH in the central Baltic Sea will fall by ca. 0.25 pH units (Kuznetsov et al. unpublished results) or up to 0.45 pH units (Omstedt et al. unpublished results). In deep water, modeling suggests increased spatial extent of anoxia, which provides increased buffering against ocean acidification. Nonetheless, modeling shows that pH in deep waters of the central Baltic Sea, which is already very low (typically ≤7.3 units), is likely to fall by at least an additional 0.1 pH units (Omstedt et al. unpublished results). This considerable variation in range overlaps the ca. 0.35 pH unit change anticipated in the open oceans under the same scenario (Cao et al. 2007; IPCC 2007), and represents a substantial shift in pH.

Research into the biological effects of ocean acidification has expanded rapidly in the last 5 years. Several reviews now show that the effects of ocean acidification vary markedly between different classes of organism, between closely related species, and between life stages within the same species (Doney et al. 2009, 2012; Byrne 2011). More detailed meta-analyses of published results have drawn divergent conclusions, although some prominent examples have been based on flawed methodologies. A recent comprehensive meta-analysis (Kroeker et al.

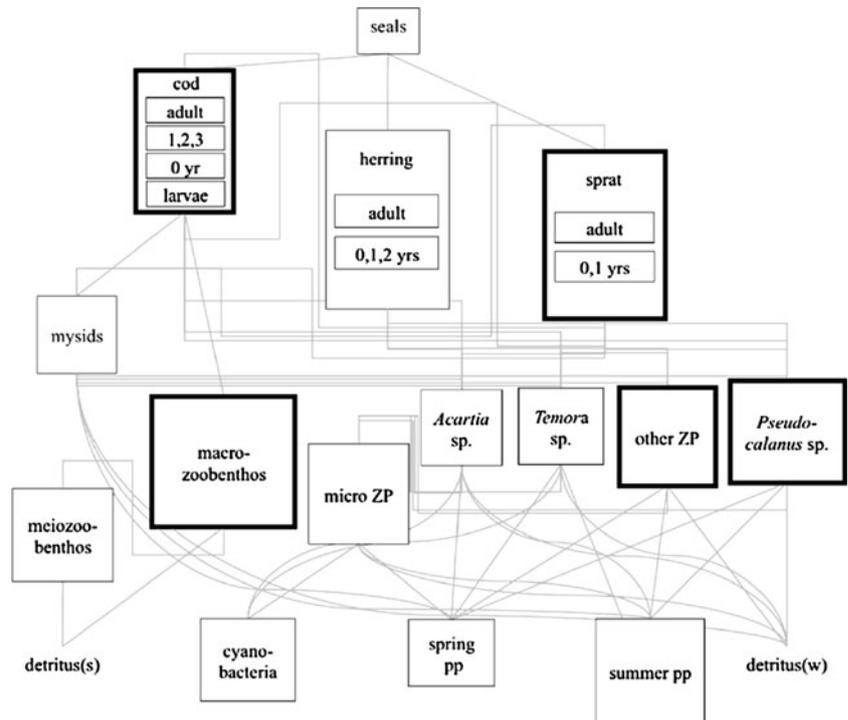
2010) concluded that overall effects on marine animals are negative, while effects on marine primary producers are generally positive. The overriding result from this analysis, however, was that substantial variation exists among the responses of different taxa and processes to acidification (e.g. corals respond differently to fishes, and effects on reproduction differ from those on survival; Kroeker et al. 2010). None of these reviews have evaluated intraspecific variation in responses to ocean acidification, and yet it is precisely this variation that provides the raw materials for the acclimation (i.e. plasticity) and adaptation (i.e. change in gene frequencies) that will be required if marine species are to survive near-future climate change.

This article reviews the available data on effects of ocean acidification on Baltic Sea marine species, focusing in particular on those taxa that modeling suggests have the greatest influence on the Baltic Sea foodweb (Niiranen et al. 2012; Fig. 1). In many cases, the effects of ocean acidification on these species have not been studied in Baltic Sea populations, and therefore summaries are provided from studies conducted elsewhere (note that the near complete absence of data precludes consideration of several ecologically important taxa such as heterotrophic bacteria, viruses and nanoflagellates). Problems involved in up-scaling from population- and species-level studies to draw ecosystem-level inferences are highlighted, and recommendations are made for future research and analyses. Where possible, differences in responses of species in the different parts of the Baltic Sea are noted (although again, available data are very few). Throughout, the general term ‘Baltic Sea’ will be used to refer to the whole Skagerrak–Kattegat–Baltic Sea system, whereas specific areas within this system will be referred to by name (e.g. “Central Baltic Sea”).

## EFFECTS OF OCEAN ACIDIFICATION ON BALTIC SEA SPECIES

Few studies have investigated the effects of ocean acidification on Baltic Sea species and populations, although many studies are now in progress and data are becoming available. Interpopulation variability in response to ocean acidification is to be expected as a consequence of adaptation to local conditions, and several authors have shown that this can be substantial in some species (Findlay et al. 2010a, b; Walther et al. 2010, 2011). It must be remembered, however, that the prevalence of such examples may not represent the prevalence of trait variability as there are likely to be biases against reporting of non-significant differences between populations, which have also been found (Havenhand and Kurihara, unpublished results). Nonetheless, given the special nature of the Baltic Sea ecosystem, caution should be used when using results

**Fig. 1** Flow chart of the BalProWeb model (Tomczak et al. 2012). *Bold boxes* indicate groups with greatest influence on model results (Niiranen et al. 2012). The effects of ocean acidification on these groups is the focus of this article (ZP zooplankton, PP phytoplankton, detritus(s) sediment detritus and detritus(w) water-column detritus). From Tomczak et al., modified



obtained from populations outside the Baltic Sea to draw inferences about the likely effects of ocean acidification on species inside the Baltic Sea system.

**Primary Producers**

The positive effects of ocean acidification on plants and algae reported from meta-analyses (Kroeker et al. 2010) are reflected in data available for Baltic Sea species. In the eelgrass *Zostera marina* from the Kattegat, ocean acidification ( $\approx 72 \mu\text{atm CO}_2$ ) increased shoot biomass in mesocosms (Eklöf et al. 2012). In studies of populations of *Z. marina* from the Pacific, photosynthesis, below-ground productivity and flowering density have all been shown to increase under elevated  $\text{CO}_2$  (Zimmerman et al. 1997; Palacios and Zimmerman 2007). At the  $p\text{CO}_2$  levels expected in the near-future ( $\leq 1000 \mu\text{atm}$ ; Cao et al. 2007; IPCC 2007) the magnitude of these responses is relatively small (Kroeker et al. 2010; Eklöf et al. 2012). Nonetheless, eelgrass growth will likely benefit from local ocean acidification (see “Ecosystem effects” below).

Benthic macroalgae are also important primary producers and ecosystem engineers in Baltic Sea ecosystems, however, very few reports in the literature detail the effects of ocean acidification on macroalgae. Most macroalgae have  $\text{CO}_2$  concentrating mechanisms (CCMs), which reduce their dependency on ambient  $\text{CO}_2$  (or  $\text{HCO}_3^-$ ) concentrations. Therefore, these taxa have been assumed to be largely insensitive to acidification (Giordano et al.

2005). There are no published results for the effects of ocean acidification on Baltic Sea macroalgae, but recent work elsewhere confirms that species with CCMs are likely to be unaffected by, or may benefit marginally from, ocean acidification (Hepburn et al. 2011). These authors also note that the responses to ocean acidification of macroalgae, such as the *Fucus* spp. common in the Baltic Sea, will be highly dependent on light intensities. Consequently, the balance between direct effects of  $p\text{CO}_2$  on macroalgal growth and the indirect effects on macroalgal photosynthesis via planktonic microalgal growth, water clarity, and hence light levels, will determine the overall outcome of ocean acidification for benthic macroalgae in the Baltic Sea.

For planktonic microalgae, early work on ecosystem-scale  $p\text{CO}_2$  effects in the Baltic Sea suggested that seasonal under-saturation of calcite<sup>1</sup> may explain the absence of calcifying coccolithophores from Baltic Sea waters (Tyrrell et al. 2008). Seasonally high  $p\text{CO}_2$  levels such as those noted by Tyrrell et al. (2008) are typically the result of remineralisation of organic material rather than a direct result of ocean acidification. But ocean acidification is likely to augment these effects (see above), and therefore the capacity for seasonal limitation of species distributions due to high  $p\text{CO}_2$  (and accordant reduced pH) is likely to increase.

<sup>1</sup> Calcite is a mineral form of  $\text{CaCO}_3$ , the saturation state of which decreases with increasing ocean acidification.

Unlike the open ocean, where substantial blooms of coccolithophores can be an important component of microalgal production, Baltic Sea phytoplankton communities are dominated by spring blooms of diatoms, and summer blooms of cyanobacteria (“blue-green algae”). Here again, relatively little work has been done on Baltic Sea species or populations. The prevalence of CCMs in phytoplankton has been suggested as a reason to expect reduced sensitivity to ocean acidification (Giordano et al. 2005; Hopkinson et al. 2011). For spring bloom species, recent results for the diatom *Skeletonema marinoi* from the Skagerrak showed that the effects of ocean acidification on growth rate differed between strains and populations but overall, growth of *S. marinoi* was not affected by increased  $p\text{CO}_2$  (750  $\mu\text{atm}$ ; Kremp et al. unpublished results). Studies from other regions have shown that diatom growth can benefit slightly from higher  $p\text{CO}_2$  levels, and when placed in competition with calcifying phytoplankton acidification led to increased competitive ability (Riebesell et al. 2007). Clearly, available evidence is scant, but on balance it seems likely that the effects of ocean acidification on the Baltic Sea spring phytoplankton bloom will be small, and perhaps positive.

The effects of ocean acidification on cyanobacteria—the major component of the summer phytoplankton bloom in the Baltic Sea—are slightly better understood. In *Nodularia spumigena*, a common toxic species in the Baltic Sea, Czerny et al. (2009) found reduced cell-division rates and increased heterocyst production as  $p\text{CO}_2$  increased over a wide range (160–730  $\mu\text{atm}$ ). More recent work found no effect of elevated  $p\text{CO}_2$  (970  $\mu\text{atm}$ ) on biovolume or specific growth rate for *N. spumigena* from the central Baltic Sea (Karlberg and Wulff 2012). Similar non-significant results were found for *Aphanizomenon* sp., another dominant Baltic Sea cyanobacteria (Karlberg and Wulff 2012). Czerny et al. (2009) suggest that the patterns they observed may be a result of adaptations to the substantial diurnal variation in pH (greater than or equal to  $\pm 1$  pH unit) that can occur in blooms of these algae in the Baltic Sea (Ploug 2008). Again, there is relatively little available evidence, but it seems probable that ocean acidification will have no, or a minor negative, effect on summer phytoplankton blooms in the Baltic Sea.

Perhaps most interestingly, the effects of ocean acidification on phytoplankton may be mediated through processes other than growth. Recent study has shown that the dinoflagellate *Alexandrium ostenfeldii*, a common (although typically not bloom forming) species in the central Baltic Sea, produces higher levels of saxitoxin under elevated  $p\text{CO}_2$  (Kremp et al. unpublished results). Similar results for other dinoflagellates have been found in other parts of the world (Fu et al. 2010). No study has yet been done on the effects of increased  $p\text{CO}_2$  on the common

toxic bloom-forming dinoflagellates in the Baltic Sea, although these findings herald the possibility that near-future climate change may increase the toxicity of at least some of the algal blooms in the Baltic Sea.

## Zooplankton

Zooplankton form a key component of the Baltic Sea food web, however, once again there are no published data from Baltic Sea species or populations with which an assessment of the likely impacts of ocean acidification can be made. Reports from the North Sea suggest that increasing  $p\text{CO}_2$  will benefit gelatinous zooplankton (Attrill et al. 2007), but these conclusions are based on environmental correlation rather than experimental observations, and therefore cannot illustrate causality (Haddock 2008). Data for copepods from other parts of the world show that high  $p\text{CO}_2$  (2000–2300  $\mu\text{atm}$ ) did not influence survival, size, development, or egg production of *Acartia steueri* (Kurihara et al. 2004) or of *Acartia tsuensis* (Kurihara and Ishimatsu 2008), but that extreme values (8000  $\mu\text{atm}$ ) did influence development (though not growth and egg production) of *Calanus finmarchicus* (Mayor et al. 2007). Modeling suggests that the copepod *Pseudocalanus* sp. and ‘other mesozooplankton’ (excluding *Pseudocalanus* sp., *Acartia* spp. and *Temora* sp.) are particularly important for the Baltic Sea food web (Fig. 1). Given that the extreme  $p\text{CO}_2$  levels used in experiments to date elicited little or no response, and that the relatively rapid generation times of copepods confers a high potential for adaptability, it seems reasonable to expect that copepods will be resilient to near-future ocean acidification ( $\leq 1000$   $\mu\text{atm}$   $\text{CO}_2$ ). Nonetheless, as noted earlier, the absence of data from Baltic Sea populations makes any conclusions about the direct effects of ocean acidification on zooplankton tentative.

## Macrozoobenthos

Several studies have investigated the effects of ocean acidification on macrozoobenthos from the Baltic Sea. Most of these come from the Kiel fjord, where summer upwelling can drive the pH down to 7.5 (Thomsen et al. 2010), well below the levels typically predicted for the open oceans by end of the century ( $\leq 1000$   $\mu\text{atm}$   $\text{CO}_2$ ; Cao et al. 2007; IPCC 2007). Here, the mussel *Mytilus edulis*, a dominant macrobenthic species in the Baltic Sea, was found to maintain shell and somatic growth rates at ca. 1400  $\mu\text{atm}$   $\text{CO}_2$  ( $\approx$  pH 7.6). *M. edulis* was also observed to recruit actively at 1000  $\mu\text{atm}$   $\text{CO}_2$  ( $\approx$  pH 7.75; Thomsen et al. 2010). Thomsen et al. (2010) suggest that this unusual ability to maintain physiological function despite substantial levels of acidification was due in part to an abundance of food, which provided the energy needed to offset

the physiological costs of maintenance and growth at such low ambient pH (Melzner et al. 2011). More sensitive early life-history stages of *M. edulis* from the Skagerrak have been shown to respond differently to ocean acidification: fertilization success *increased* at reduced pH (induced by high  $p\text{CO}_2$ ) whereas subsequent larval shell growth was negatively affected, albeit only slightly (Renborg and Havenhand, unpublished results). Similar small negative effects on larval shell growth have also been reported in populations of *M. edulis* from the North Sea (Gazeau et al. 2010; Bechmann et al. 2011), and in related *Mytilus* species around the world (Kurihara et al. 2009; Gaylord et al. 2011; Sunday et al. 2011). Early reports of the effects of ocean acidification on shell growth in adult *M. edulis* showed negative impacts (Gazeau et al. 2007), a result that contrasts with those of Melzner and co-workers in the Kiel fjord (Thomsen et al. 2010; Melzner et al. 2011), although the latter might be expected to be a result of local adaptation to seasonally low pH—especially at such extreme levels (Melzner et al. 2009b; Thomsen et al. 2010).

Together with *M. edulis*, the clam *Macoma balthica* dominates the seafloor of much of the central and northern Baltic Sea. These two species comprise the overwhelming majority of the ‘macrozoobenthos’ category identified as having a marked effect on output from ecosystem models of the central Baltic Sea (Niiranen et al. 2012). Unfortunately, there are no available published data on the effects of ocean acidification on *Macoma* species, and therefore it is not possible to draw meaningful conclusions on the likely sensitivity of this key component of the Baltic Sea food web. It is clear, however, that tolerance of *M. edulis* to ocean acidification increases with available food ration (Melzner et al. 2011). Increased ocean acidification represents an additional stress on marine organisms that can cause corresponding increases in maintenance and physiological costs (Pörtner et al. 2004; Pörtner 2008). This ‘acidification stress’ will operate in concert with other stressors that limit the distribution and function of species—particularly salinity, which is a major determinant of mussel size and distribution in the Baltic Sea (Tedengren and Kautsky 1987; Westerbom et al. 2008). Food-dependent tolerance to ocean acidification has not only been observed in the mussel (Melzner et al. 2011), but also in barnacle, *Balanus improvisus*, from the Baltic Sea (Pansch, unpublished results), as well as in other species elsewhere. Consequently, the availability of adequate energy reserves in the form of food may determine tolerance to ocean acidification in many species of macrozoobenthos. This emphasises the need for multifactor experiments on multiple trophic levels (see “Ecosystem effects” below).

In summary, available evidence suggests changes in macrozoobenthos as a result of future ocean acidification will be small, however, there is considerable uncertainty in

this assessment and many more data from additional populations throughout the Baltic Sea are required.

## Fish

Cod, herring and sprat dominate fisheries in the Baltic Sea (Sparholt 1991; Heikinheimo 2011) and are key components of Baltic Sea food web models (Harvey et al. 2003). These food web models are highly sensitive to changes in cod and sprat abundances (Niiranen et al. 2012), suggesting that the effects of ocean acidification on these species may have more far-reaching consequences than for other components of the Baltic Sea food web. The rapidly growing body of data reporting the effects of ocean acidification on cod and herring species shows that these species appear to be relatively robust to considerable  $p\text{CO}_2$ -mediated changes in pH. For example, sperm motility in Baltic Sea cod has been shown to be insensitive to moderate levels of acidification ( $\leq 1360 \mu\text{atm CO}_2 \approx \text{pH } 7.55$ ; Frommel et al. 2010), leading these authors to conclude that fertilization success in cod would also be unaffected by near-future acidification. Similarly, fertilization, embryogenesis, hatching success and larval growth of Baltic Sea herring were unaffected by  $p\text{CO}_2$  up to  $4600 \mu\text{atm}$  (Franke and Clemmesen 2011). At higher  $p\text{CO}_2$ , RNA/DNA ratios were reduced suggesting that growth may be impaired, however, there were no statistically significant effects on any of the variables measured at  $p\text{CO}_2$ s representative of IPCC scenarios for the years 2100 and 2200 (Franke and Clemmesen 2011). As for the *M. edulis* example discussed earlier, these herring data come from the Kiel fjord, and Franke and Clemmesen (2011) note that their results may reflect adaptation of this population to increasing  $p\text{CO}_2$  in the surface waters during larval development.

Data from Norwegian cod populations show that tolerance to ocean acidification declines gradually early larval stages, leading to some organ damage in later larvae and juveniles (Frommel et al. 2012). Melzner et al. (2009a), also using Norwegian cod, exposed juveniles reared under normal  $p\text{CO}_2$  conditions to elevated ( $3000 \mu\text{atm}$ )  $p\text{CO}_2$  for several months, and found no effect on metabolic rates and critical swimming speeds (Melzner et al. 2009a). The consequences of the changes observed by Frommel et al. (2012) on swimming performance of later juveniles are not known. Equivalent data for herring and sprat are currently lacking; however, on balance, it seems likely that the effects of ocean acidification on cod and herring will be small, although negative impacts may be experienced in later larval stages. Once again, experimental data from Baltic Sea populations are needed in order to reduce the uncertainties around this estimate.

In a much broader ecosystem context, dynamic climate envelope modeling of the effects of climate change on

global fisheries suggests that an overall lack of (or low) sensitivity to ocean acidification (as observed here for most species and life stages) will result in no net change to maximum catch potential in Baltic Sea fisheries by 2050 (Cheung et al. 2011). However, Cheung et al. (2011) caution that if commercially important Baltic Sea fish species are in fact sensitive to ocean acidification, then maximum catch potential may fall by up to 30 %.

### Ecosystem Effects

Responses of ecosystems to ocean acidification will be influenced by multiple abiotic factors (notably temperature and salinity) as well as by interactions between species. The effects of ocean acidification on competition, predation and mutualisms are only beginning to be understood, and examples are few. At the time of going to press there are no published studies that have investigated the effects of multiple climate change variables on multiple trophic levels in the Baltic Sea region, although Eklöf et al. (2012) investigated the effects of ocean acidification and warming on seagrass mesocosms in the Skagerrak. They found substantial effects of warming, and small positive effects of acidification on seagrasses and macroalgae, and neutral effects on grazers in the system. Importantly, these impacts were context dependent: acidification amplified the effects of warming, but only in the absence of grazers (Eklöf et al. 2012). This result could not have been predicted from single-species studies, highlighting the need for multifactor multi-species approaches. Equivalent patterns can be found in literature on the effects of ocean warming (e.g. Kordas et al. 2011), and therefore it seems clear that changing abiotic factors will alter competitive, and trophic, interactions of key functional groups (Hepburn et al. 2011). Understanding these changes will be fundamental to understanding how ecosystems will respond in a high CO<sub>2</sub> ocean.

### CONCLUSIONS

The responses of marine organisms to ocean acidification vary markedly between populations, between species and between life-history stages. Consequently, drawing conclusions as to the likely effects of ocean acidification is problematic at best. For Baltic Sea ecosystems conclusions are severely constrained by the lack of experimental and observational data for many functional groups. Available data from experiments that have used Baltic Sea populations show that many key taxa in the food web of the Baltic Sea are generally tolerant of the pCO<sub>2</sub>-mediated pH excursions expected in the coming century, or respond only slightly. Exceptions to this pattern are larval stages of mussels and cod, which may experience biologically

significant negative impacts. There is, however, a paucity of data and future studies may find more significant effects.

In attempting to place these findings in context it must be remembered that ocean acidification will operate in concert with other environmental variables to change competitive, predator and mutualistic interactions among species. This will in turn reshape the fitness and selection landscapes in ways that the single-factor, single-species experiments that have dominated the ocean acidification literature cannot inform. How Baltic Sea species respond to these changed landscapes will depend on their capacity for acclimation and adaptation (plasticity and genetic change, respectively) in response to *all* the stresses placed upon them, not solely to ocean acidification. To use a colloquial analogy, when two friends out walking in the forest were suddenly confronted with a charging bear and one of them started to run, the other shouted “Why run? The bear can outrun us both!” His friend responded “I don’t need to run faster than the bear, I only need to run faster than you!” It is not the absolute response to ocean acidification, but the balance between fitness costs and fitness benefits of ocean acidification that will determine winners and losers in a future high CO<sub>2</sub> Baltic Sea. At present, there are too few data to be able to state with any degree of certainty which taxa will fall into which category.

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